

## DRUG CHEMISTRY

CHD - 361 : Chemistry of Heterocycles and Drug Synthesis  
(2013 Pattern) (Credit System) (Semester - III)

Time : 3 Hours]

[Max. Marks : 50

Instructions to the candidates:

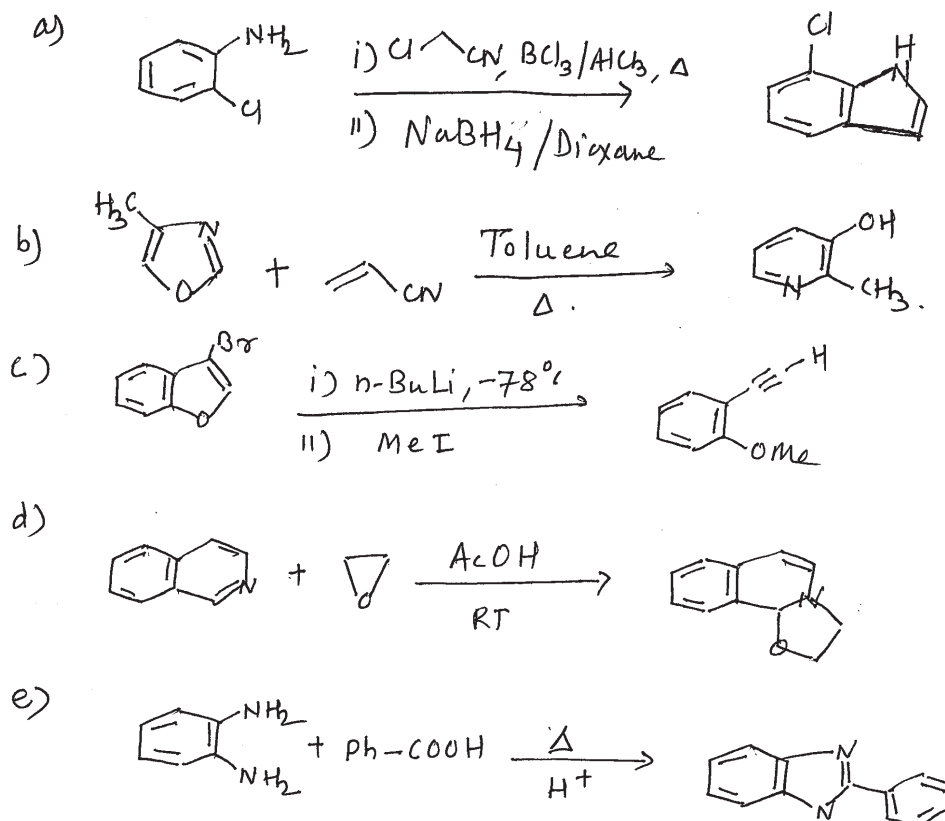
- 1) All questions are compulsory.
- 2) Figures to the right indicates full marks.
- 3) Answers to the two sections should be written in separate answer books.

SECTION - I

Q1) Explain any four of the following : [8]

- a) 4-Chloropyridine easily undergoes hydrolysis in warm water.
- b) Explain the synthesis of furan from pentose.
- c) Pyrazole has higher boiling point than iso-oxazole.
- d) Coumarin undergo cycloaddition reaction.
- e) 2-Aminoquinoline on diazotization give 2-quinolone.

Q2) Suggest the suitable mechanism for any four of the following : [8]

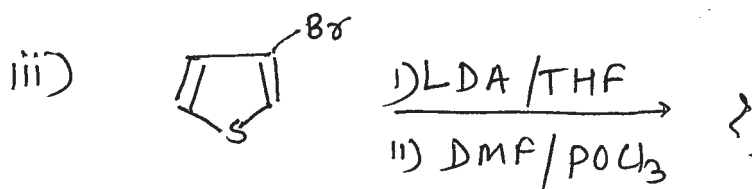
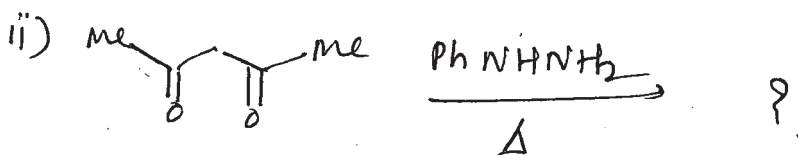
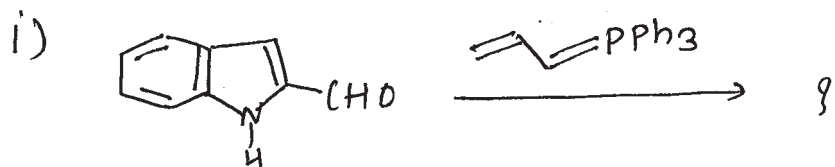


P.T.O.

Q3) a) Write short notes on any two of the following : [4]

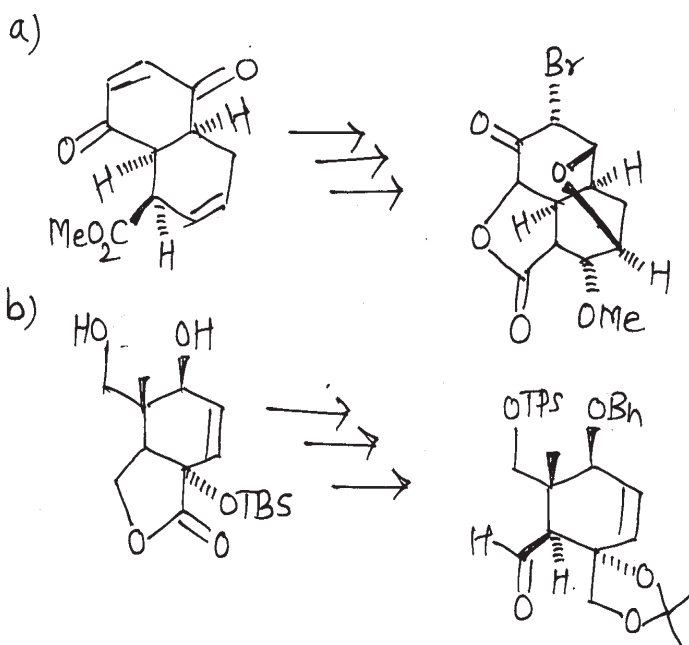
- i) Tschitschibabin Reaction.
- ii) Medlung Indole Synthesis.
- iii) Paal-Knorr synthesis of Pyrrole.

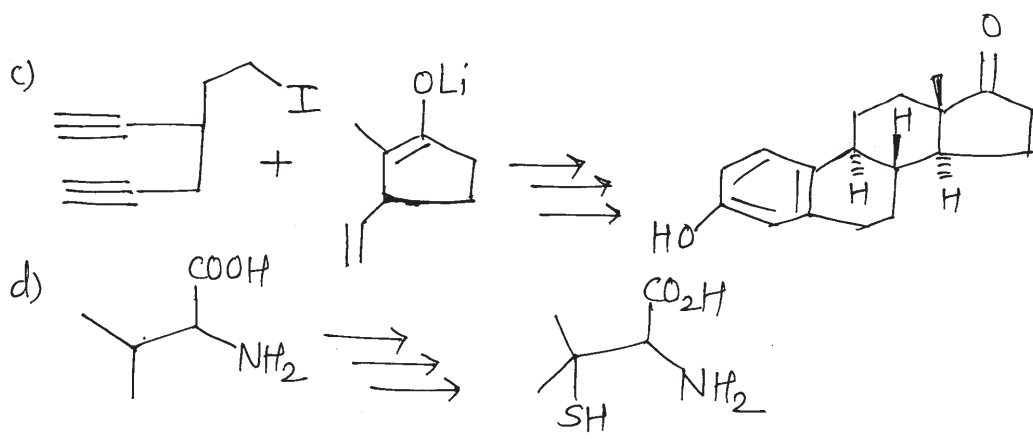
b) Predict the products for any two of the following : [5]



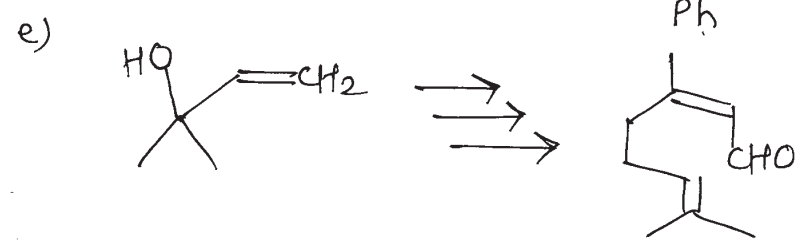
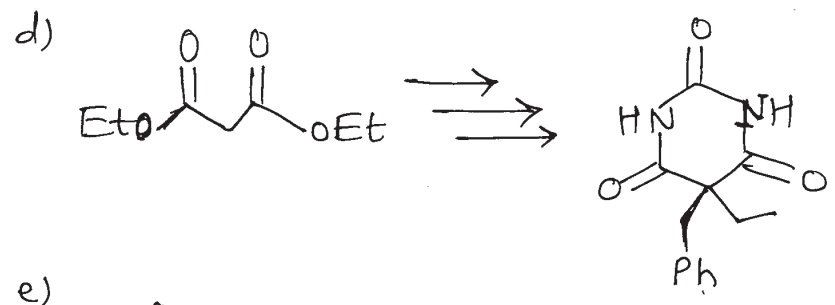
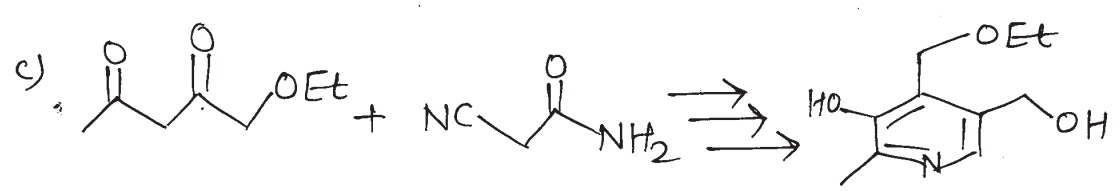
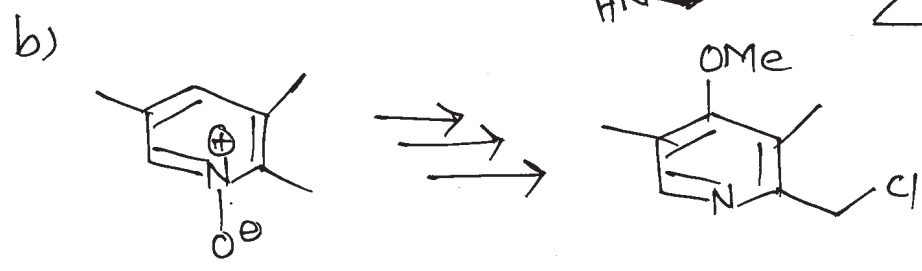
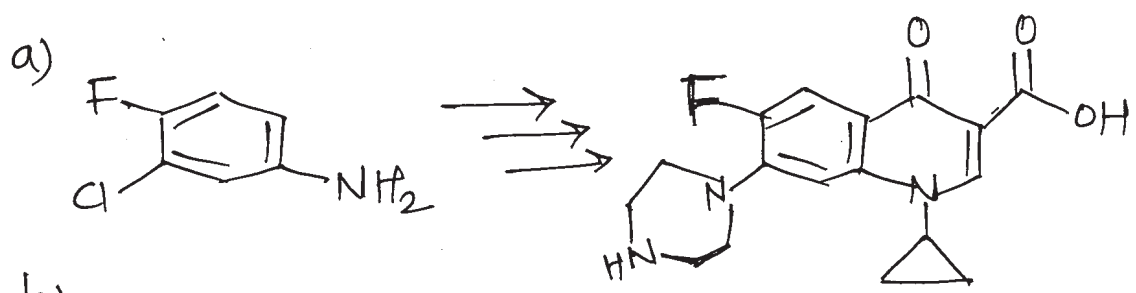
### SECTION - II

Q4) Discuss the steps involved in the synthesis of following molecules. Explain the stereochemistry and mechanism involved (any three) [9]





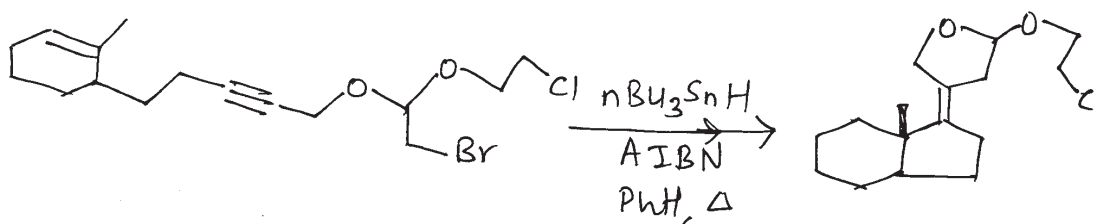
Q5) Discuss the steps involved in the synthesis of following drug molecules from the precursor shown (any four) : [10]



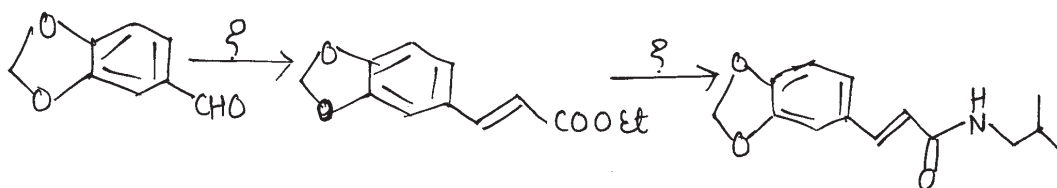
Q6) Answer any two of the following :

[6]

a) Explain the mechanism in a given transformation.



b) Insert the missing reagents in the given sequence of reaction. Explain the steps with mechanism.



c) Explain any one of the following :

- Shapiro reaction.
- Gabriel synthesis.



Total No. of Questions :6]

SEAT No. :

**P1251**

**[5432]-3002**

[Total No. of Pages : 7

**M.Sc.**

**DRUG CHEMISTRY**

**CHD-362: Advanced Analytical Methods**

**(2013 Pattern) (Semester-III)**

*Time : 3 Hours]*

*[Max. Marks : 50*

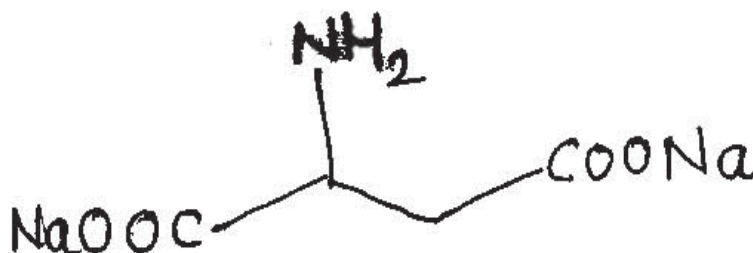
*Instructions to the candidates:*

- 1) *All question are compulsory.*
- 2) *Answers to the two sections should be written in separate answer book.*
- 3) *Figures to the right indicate full marks.*

**SECTION-I**

**Q1) a)** Answer the following. (any three) **[6]**

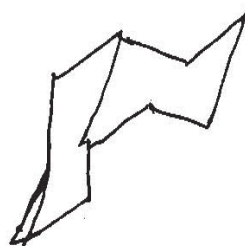
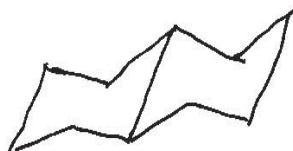
- i) Explain the splitting pattern and intensity of signals in 1, 4 - dioxane- $d_8$  in  $^{13}C$  NMR.
- ii) O-Dichlorobenzene shows peaks at 96, 98 and 100 in the ratio 9:6:1 in its MS. Explain.
- iii) Sodium Asparate in  $D_2O$  show signals in  $^1H$  NMR at 3.5 (dd,  $J=10$  & 4 Hz), 2.4 (dd,  $J=15$  & 4Hz), 2.2 (dd,  $J=15$  & 10 Hz). Explain.



- iv) DEPT is more helpful than APT in CMR. Explain.
- b) Distinguish between the following pairs using the indicated spectral methods. (any two) **[3]**

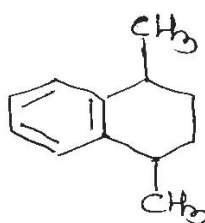
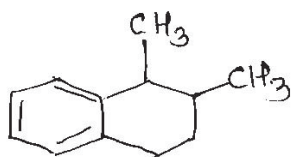
*P.T.O.*

i)



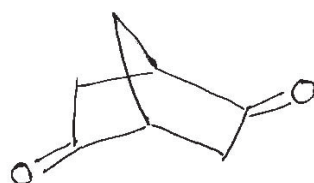
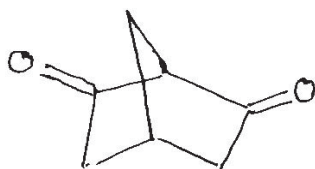
PMR

ii)



Mass

iii)



CMR

**Q2)** Using the given spectral information, deduce the structure of the following.  
(any four). **[10]**

a) M.F:  $C_9H_{12}O_3S$

PMR: 1.3 (t,  $J=6\text{Hz}$ , 3H), 2.45 (s, 3H),  
4.1 (q,  $J=6\text{Hz}$ , 2H), 7.45 (d,  $J=8\text{Hz}$ , 2H)  
8.8 (dd,  $J=8\text{Hz}$ , 2H).

b) M.F:  $C_{10}H_{10}O_3$

IR : 1680, 1602  $\text{cm}^{-1}$

PMR: 3.1 (t,  $J=6\text{Hz}$ , 2H), 3.9 (s, 3H),  
4.5 (t,  $J=6\text{Hz}$ , 2H), 6.7 (d,  $J=2\text{Hz}$ , 1H)  
6.9 (dd,  $J=8 \text{ \& } 2 \text{ Hz}$ , 1H), 8.05 (d,  $J=8 \text{ Hz}$ , 1H)

c) M.F:  $C_8H_9NO$

CMR: 161 (d), 142 (s), 129 (d, str.)  
125 (d, str.), 121 (d), 31 (q)

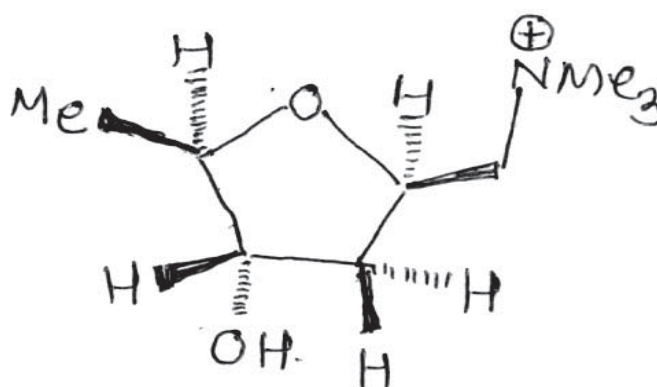
- d) M.F:  $C_5 H_{10} O_3$   
 IR:  $1728 \text{ Cm}^{-1}$   
 PMR: 2.1 (s, 3H), 3.35 (S, 6H), 4.6 (S, 1H)  
 CMR: 25, 55, 104, 204  
 DEPT90: 25, 55, 104, no peak , 104 up  
 DEPT135:204 no peak, 25, 55, 104 up.
- e) IR:  $1715 \text{ cm}^{-1}$   
 Mass: 128 ( $M^+$ , 3%), 85 (10%), 72 (40%), 43 (100%)

**Q3)** Write notes on the following. (any three). [6]

- Nuclear overhauser Effect.
- Chemical and magnetic shift equivalence in NMR
- Attached proton Test

### SECTION-II

**Q4) a)** Assign the chemical shifts to various protons in the given structure. Comment on the observed coupling constants and double resonance experiments. [5]



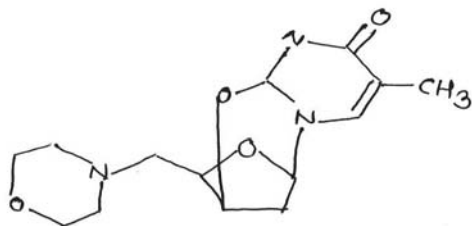
1.6 (d,  $J=6.5 \text{ Hz}$ , 3H), 1.86 (ddd,  $J=12.5, 9.5 \text{ \& } 5.5 \text{ Hz}$ , 1H), 2.02 (ddd,  $J=12.5, 6 \text{ \& } 2 \text{ Hz}$ , 1H), 3.36 (S, 9H), 3.54 (dd,  $J=13 \text{ \& } 9 \text{ Hz}$ , 1H), 3.74 (dd,  $J=13 \text{ \& } 1 \text{ Hz}$ , 1H), 3.94 (dq,  $J=6.5 \text{ \& } 2.5 \text{ Hz}$ , 1H), 4.03 (m, 1H), 4.30 (d,  $J=3.5 \text{ Hz}$ , 1H, exch.) 4.68 (m, 1H)

Spin- decoupling Expt.

- i) Irradiation at  $\delta$  3.92 Changes  $\delta$  4.03  $\rightarrow$  dd, 9.5 & 2 Hz  
ii) Irradiation at 4.68 Changes  $\delta$  2.02  $\rightarrow$  dd, 12.5 & 2 Hz  
Changes  $\delta$  1.86  $\rightarrow$  dd, 12.5 & 9.5 Hz

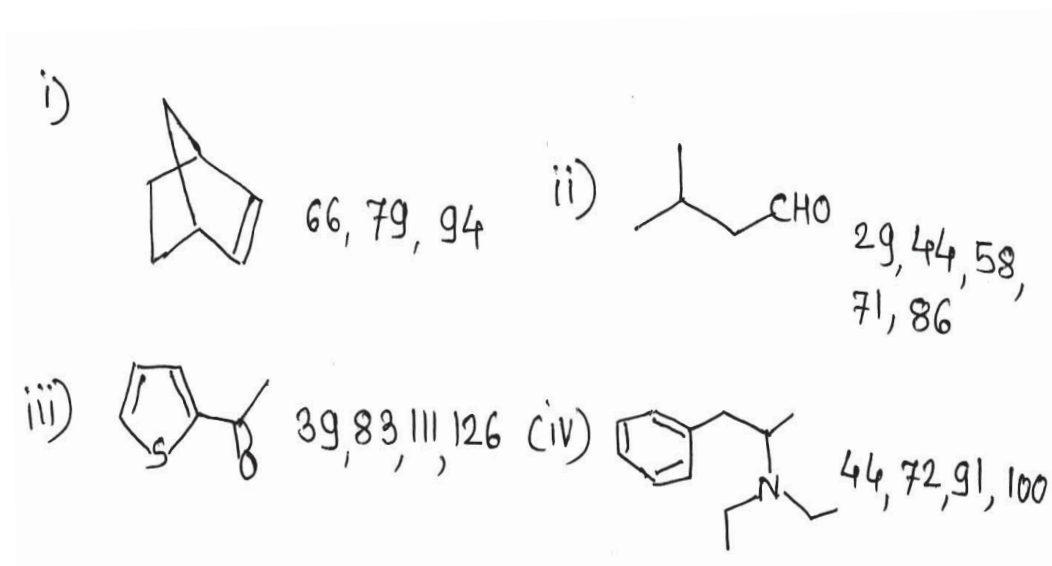
In NOe Expt. Irradiation at  $\delta$  1.6 increases the signal at  $\delta$  1.86 by 7 %

- b) Assign the chemical shift to various carbons atoms. [3]



175.9 (s), 156.91 (s), 140 (d), 118.8 (s), 89.4 (d), 84 (d), 80.6 (d), 67.2(t),  
59.1(t), 54.3 (t), 33.8 (t), 13.6 (q)

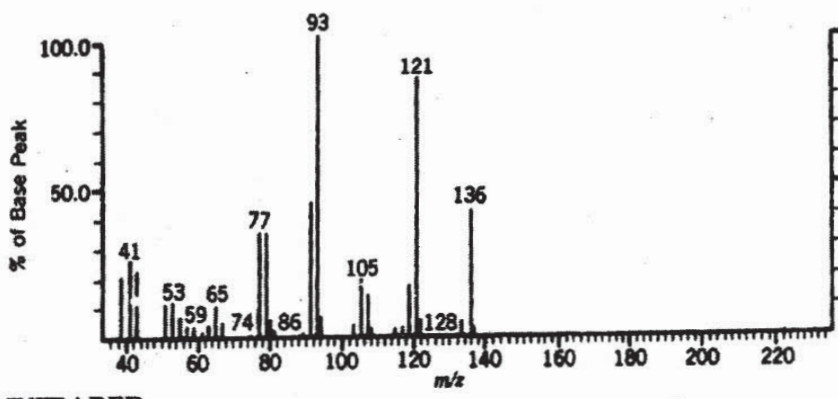
- Q5) a) Write the genesis of the indicated ion for any three of the following. [6]



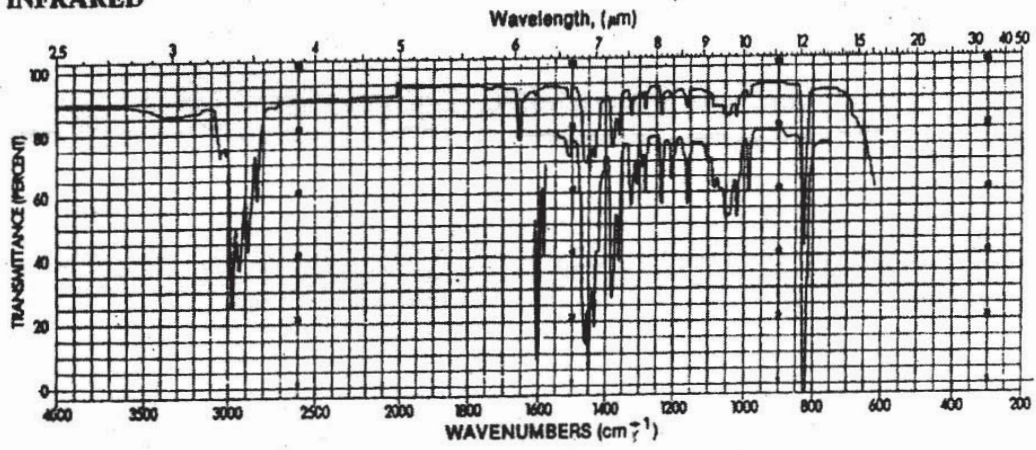
- b) Differentiate between stokes and Anti-stokes Raman spectrum. [2]

- Q6) Deduce the structure of the compound whose spectral information is given on the next page [9]

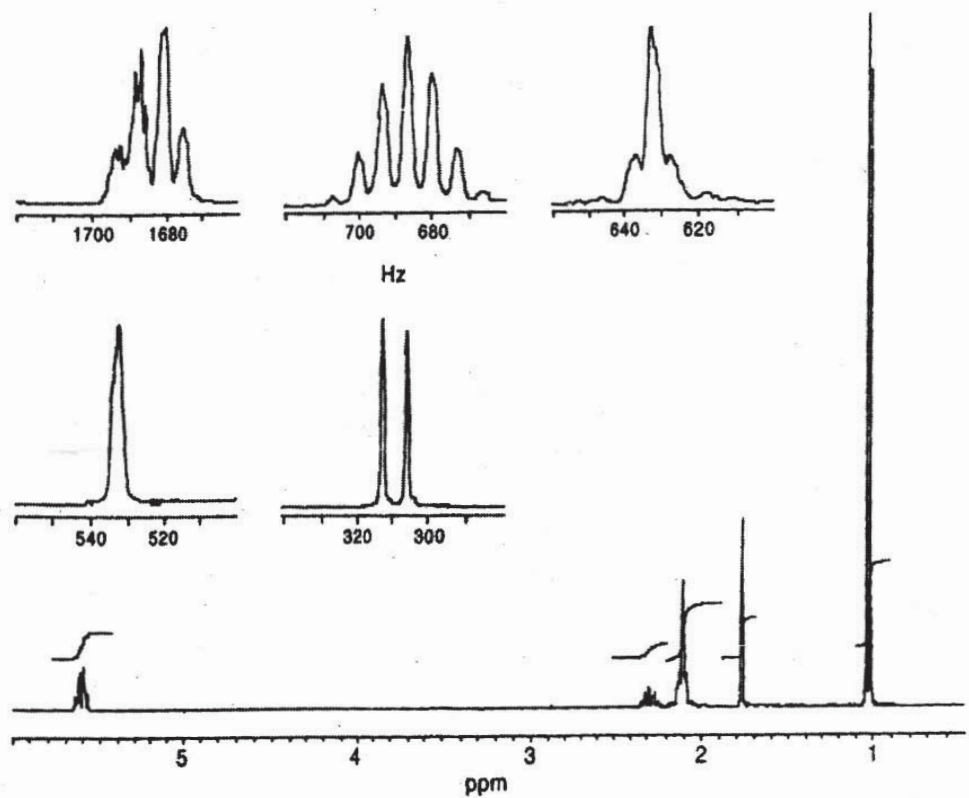


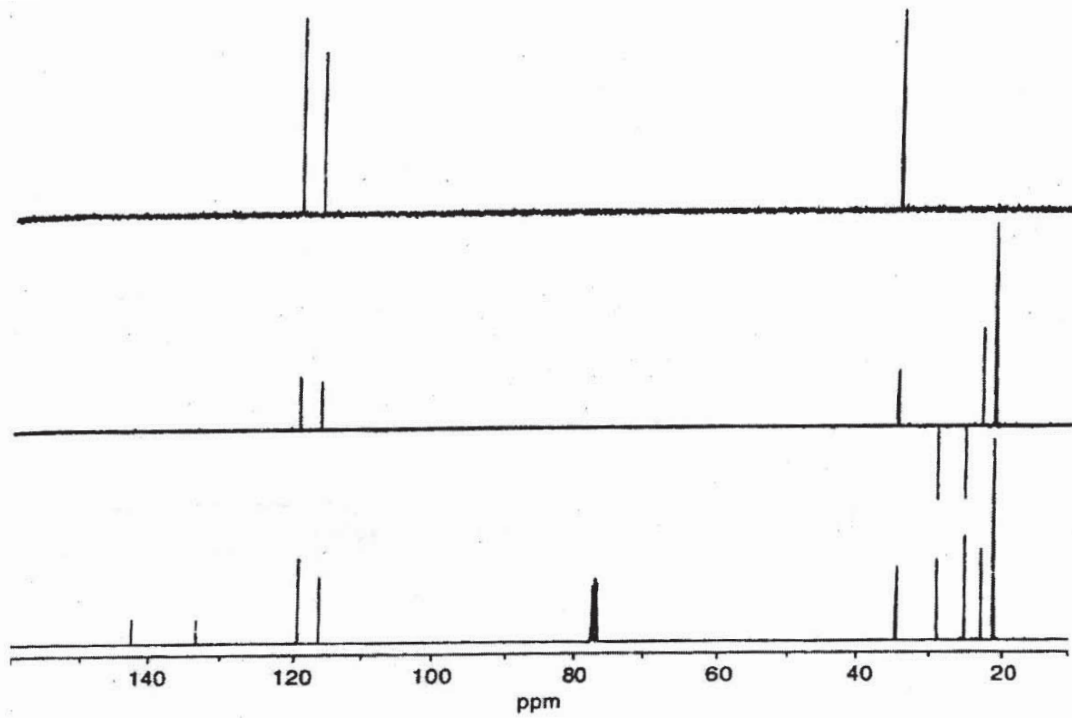


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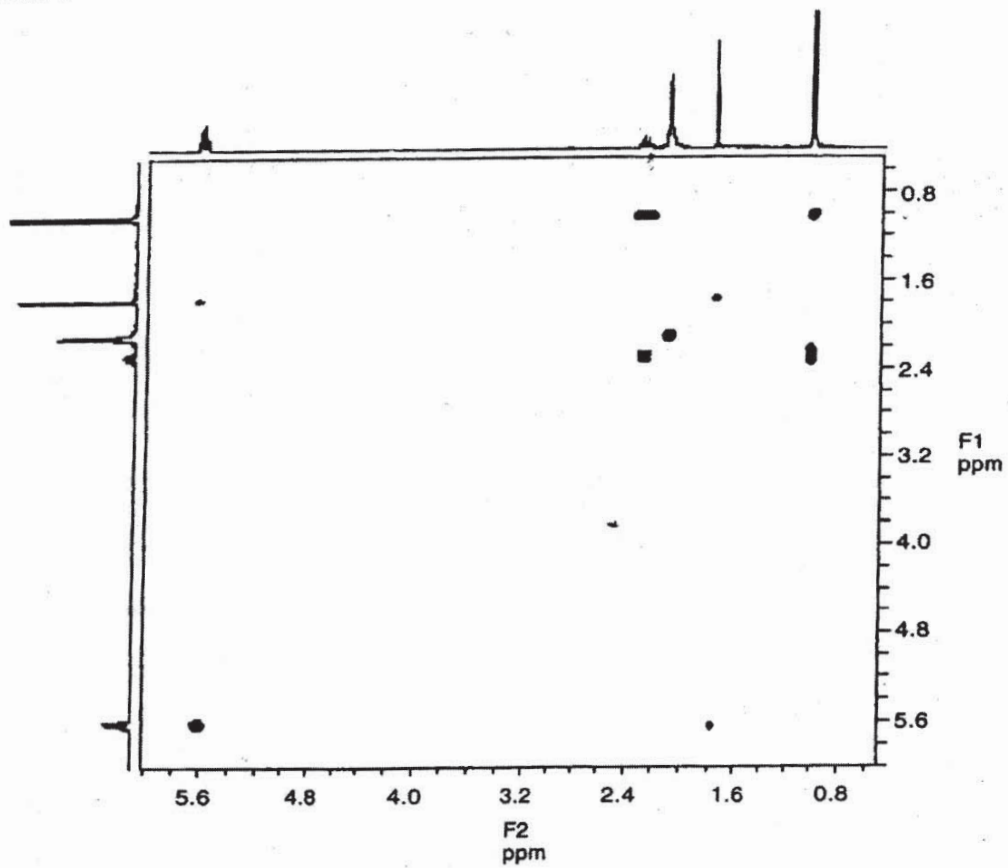


**<sup>1</sup>H NMR**

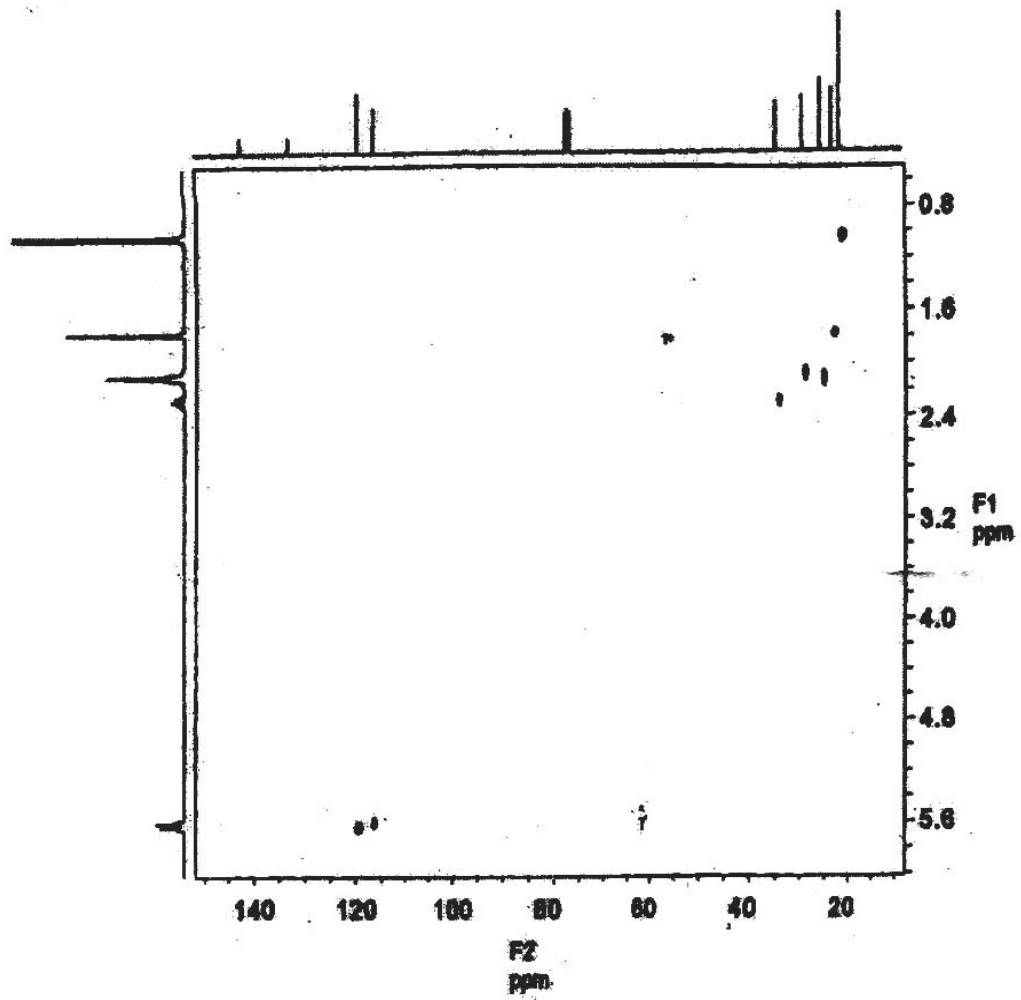




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HETCOR



Total No. of Questions : 6]

SEAT No. :

[Total No. of Pages : 2

**P1252**

**[5432]-3003**

**M.Sc.**

**DRUG CHEMISTRY**

**CHD - 363 : Microbiology, Immunology & Drug Discovery &  
Development**

**(2013 Pattern) (Credot System) (Semester - III)**

*Time : 3 Hours]*

*[Max. Marks : 50*

*Instructions to the candidates:*

- 1) *All questions are compulsory.*
- 2) *Answers to the two sections should be written in separate answer book.*
- 3) *Figures to right indicate full marks.*

**SECTION - I**

**Q1)** Attempt any three of the following: **[12]**

- a) Describe various phases of Bacterial growth curve.
- b) State the parts of a typical fermentor & give its use.
- c) Describe any one method of antimicrobial assay.
- d) Short note on: Classification of microbes.
- e) What is strain improvement of microbes. Describe any one method of the same.

**Q2)** Attempt any three of the following: **[9]**

- a) What are antibodies? Describe the structure of a typical antibody.
- b) Briefly describe the cells involved in Immune system.
- c) What are Immunosuppressants and immunomodulators, state their clinical significance.
- d) Short note on: Immunoelectrophoresis / Double diffusion technique (any one).
- e) Describe either type I or type II Hypersensitivity.

**Q3)** Explain any four of the following terms: **[4]**

- a) Immunization
- b) Cytokines
- c) Efficacy
- d) Potency
- e) Drug target
- f) MIC

**P.T.O.**

## SECTION - II

**Q4)** Answer any three of the following: **[12]**

- a) Explain the following.
  - i) Chronic toxicity studies.
  - ii) Genetotoxicity studies.
- b) What is solid dosage forms? Explain it's types with examples.
- c) What is lead? Discuss the need for lead development.
- d) Explain all the phases involved in clinical trials?
- e) Define pharmacokinetics of drug action. What factors affect the bioavailability of a drug.

**Q5)** Answer any two of the following: **[8]**

- a) What is patent? Make a comment on patentable inventions.
- b) Define following with example.
  - i) Bioavailability
  - ii) Bioequivalence
- c) What is bioassay? Write the necessity and type of Bioassay.

**Q6)** Answer any two of the following: **[5]**

- a) Ayurveda system of medicine.
- b) Pharmacopeia.
- c) Pharmacophore Identification.



Total No. of Questions : 6]

SEAT No. :

P1253

[Total No. of Pages : 3

[5432]-3004

M.Sc.

**DRUG CHEMISTRY**

**CHD - 364 : Stereochemistry Asymmetric Synthesis and Pericyclic Reactions**

**(2013 Pattern) (Semester - III)**

*Time : 3 Hours]*

*[Max. Marks : 50*

*Instructions to the candidates:*

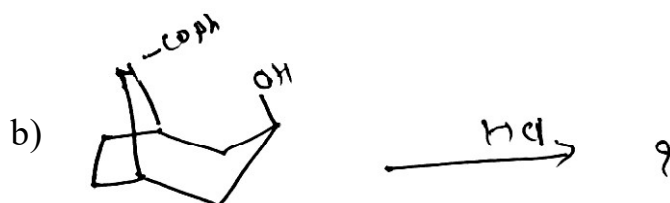
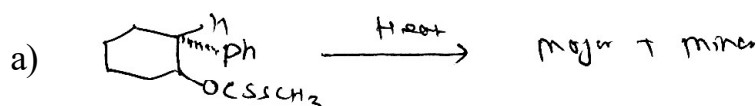
- 1) *All questions are compulsory.*
- 2) *Figures to the right indicate full marks.*
- 3) *Answers to the two sections should be written in separate answerbooks.*

**SECTION - I**

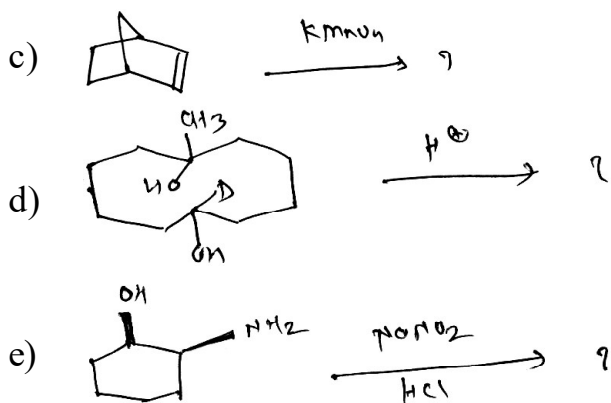
**Q1)** Answer any four of the following : **[10]**

- a) Bromo camphor fails to undergo dehydrobromination on treatment with base. Explain.
- b) Explain why chair-boat interconversion is more facile in cyclohexanene than in cyclohexane.
- c) 1, 2, 2, 6, 6 - Penta methyl-4-hydroxy-4-penal piperidine more stable in boat conformation. Explain.
- d) Draw the structures of Cis-t-trans and Cis-t-cis isomers of perydrophenanthrene. Comment on their stability & optical activity.
- e) Cis 4-t-butyl Cyclohexane carboxylic acid lactonise, whereas trans isomer do not. Explain.

**Q2)** Predict the product/s and explain the stereochemical principles involved. (any four): **[10]**



**P.T.O.**



**Q3)** Write short notes on (any two) :

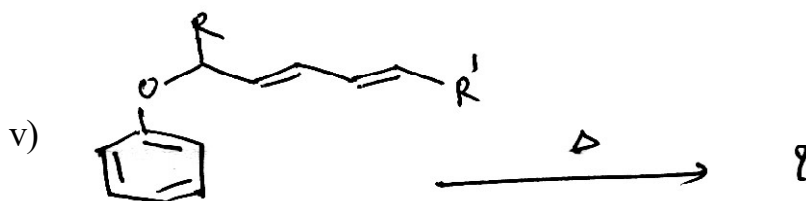
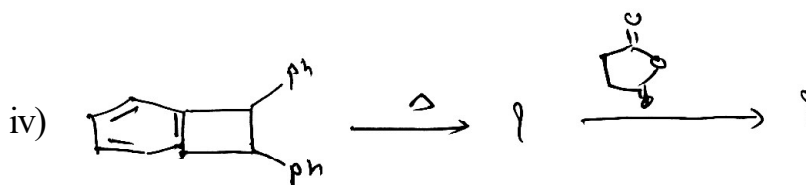
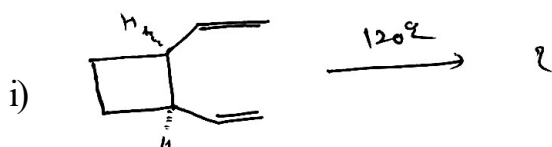
[5]

- Bredt's rule in Bridge compounds.
- Trans annular interactions
- Von Auwer's skita rule

### SECTION - II

**Q4) a)** Draw correlation diagram of CON rotatry ring opening of 1,3-cyclohexa diene to 1,3,5, hexa triene predict the allowed process. [4]

b) Predict the product/s in any four of the followings as justify your answer (any four) : [6]



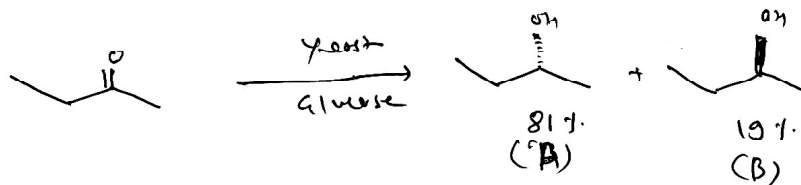
Q5) a) Attempt the following :

[4]

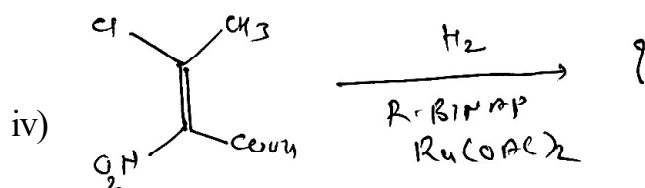
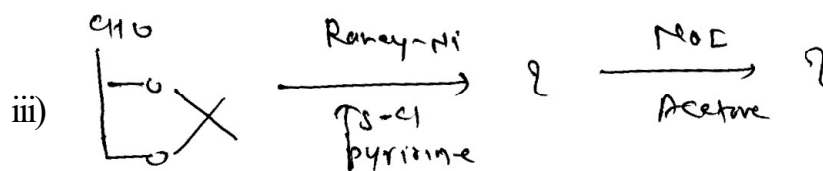
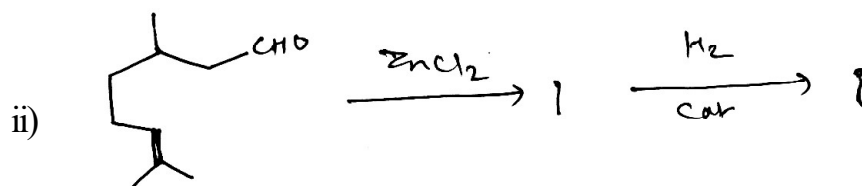
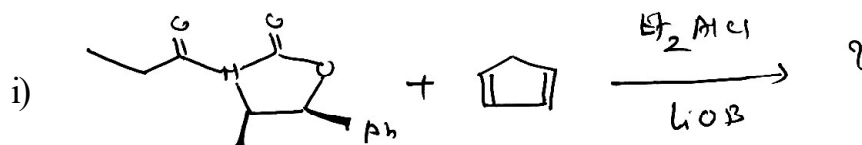


with the help of above reaction explain Felkin-Anh model.

ii) Calculate the diastereomeric excess in the following reaction Products.



b) Complete the following reactions give the mechanism involved (any three) [6]



Q6) Write short notes (any two) :

[5]

- Chiral borohydride
- Asymmetric epoxidation
- General methods of asymmetric synthesis





Total No. of Questions : 6]

SEAT No. :

P1254

[5432]-4001

[Total No. of Pages : 4

M.Sc. - II

**DRUG CHEMISTRY**

**CHD - 461 : Advanced Organic Synthesis, Principles and Strategies  
(2013 Pattern) (Semester - IV)**

Time : 3 Hours]

[Max. Marks : 50

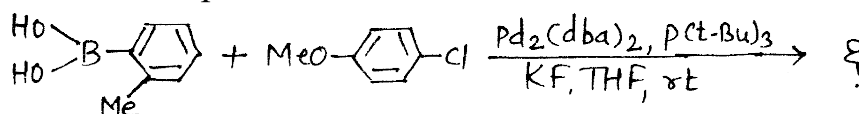
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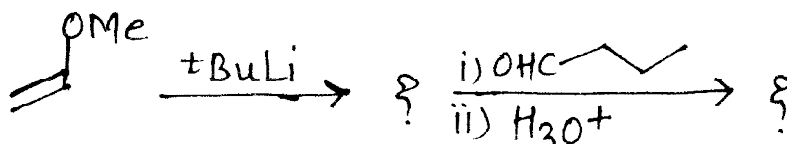
**SECTION - I**

Q1) a) Answer any three of the following: [6]

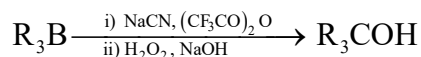
i) Predict the product



ii) Complete the following transformation.

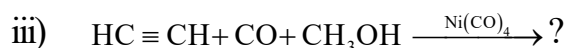
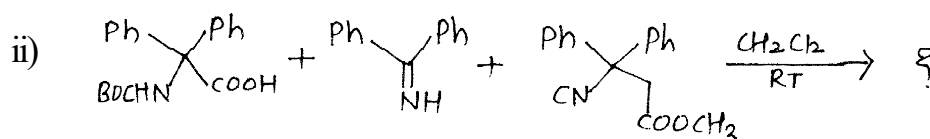
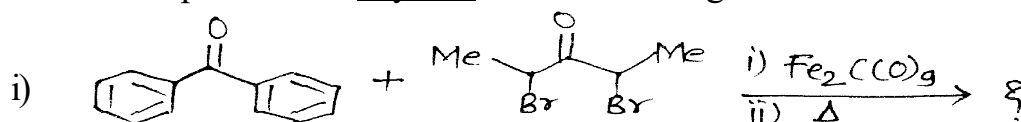


iii) Suggest the mechanism.



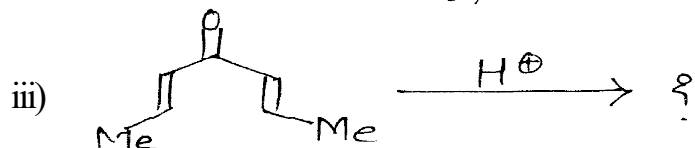
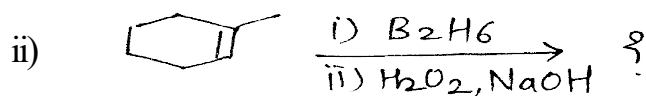
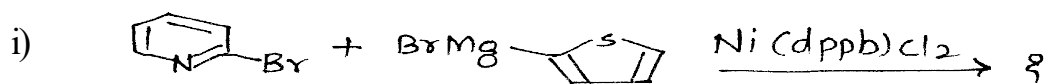
iv) Non-terminal alkenes can be converted to terminal alkenes using hydroboration reaction.

b) Predict the product for any two of the following reactions. [4]

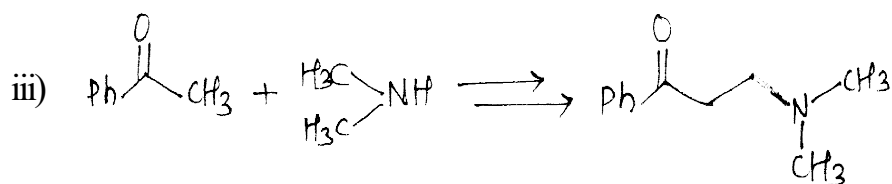
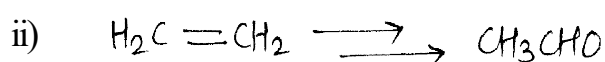
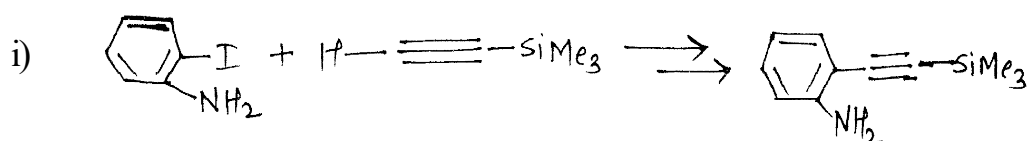


P.T.O.

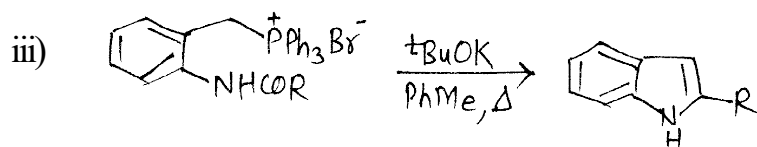
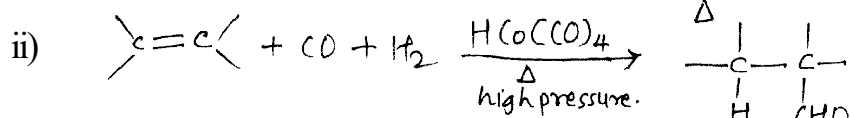
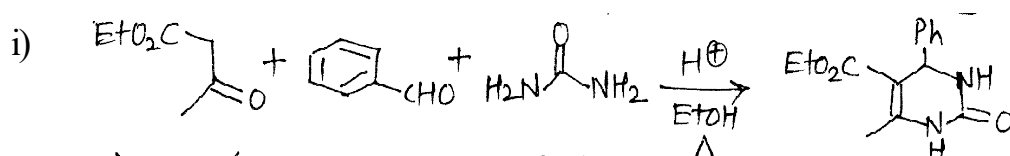
**Q2) a)** Predict the product and suggest a suitable mechanism for its formation. (any two) [4]



**b)** Carry out the following conversions and justify your answer. (any two) [4]



**Q3) a)** Explain the mechanism for any two of the following. [4]



**b)** Write short note on any two: [3]

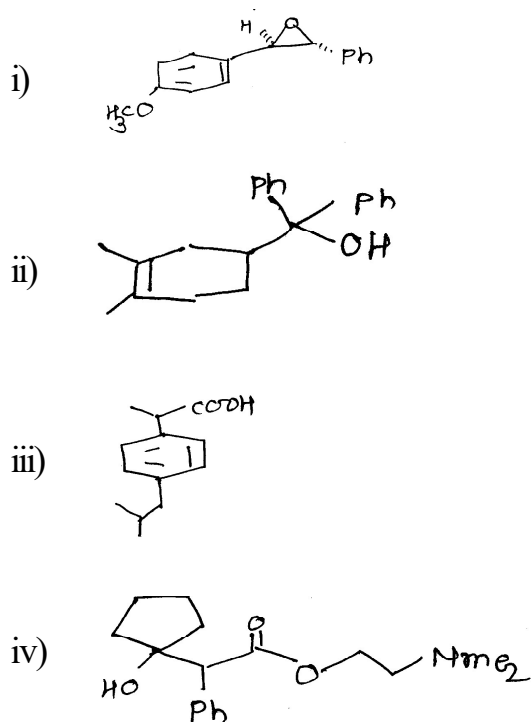
i) Sharpless azides cycloaddition.

ii) Bergmann cyclization.

iii) Stille coupling.

## SECTION - II

**Q4)** Using retrosynthetic analysis suggest a suitable method to synthesize any three of the following. [9]



**Q5) a)** Answer any two of the following. [4]

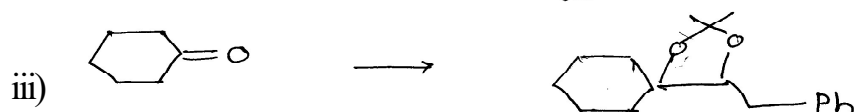
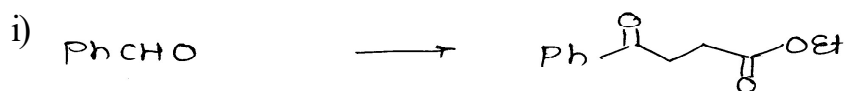
i) Give one reaction with a reagent for each synthon given below.



ii) Discuss three methods for the carbon-carbon double bond formation.

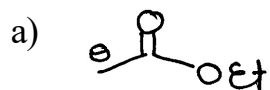
iii) Benzyl oxy carbonyl protection is preferred over benzyl group for protection of  $-\text{NH}_2$  group of amino acid in peptide synthesis.

b) Complete the following transformation. (any two) [4]

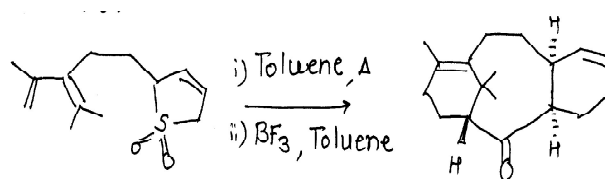


**Q6) a)** Answer any two of the following: [4]

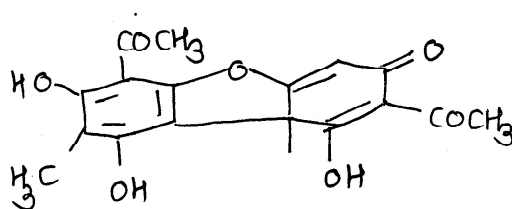
- i) Atom economy in Green Chemistry.
- ii) MOM ether protection is preferred over methyl ether protection for protection of hydroxyl group.
- iii) Give one reaction with a reagent for each synthon given below.



- b) i) Explain the advantages of cascade reaction. Write the steps involved in the following reaction. [2]



- ii) Explain the biomimetic approach to the retrosynthesis of the following [2]



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Total No. of Questions : 6]

SEAT No. :

**P1255**

**[5432]-4002**

[Total No. of Pages : 3

**M.Sc. - II**

**DRUG CHEMISTRY**

**CHD - 462 : Advanced Medicinal Chemistry**

**(2013 Pattern) (Semester - IV)**

*Time : 3 Hours]*

*[Max. Marks : 50*

*Instructions to the candidates:*

- 1) *All questions are compulsory.*
- 2) *Answers to the two sections should be written in separate answer books.*
- 3) *Figures to the right side indicate full marks.*

**SECTION - I**

**Q1)** Answer any three of the following. **[9]**

- a) Give a brief account of macrolide antibiotics. Explain the development of Clarithromycin, Roxithromycin and Azithromycin from Erythromycin.
- b) Discuss in brief the discovery of carbapenems. Discuss their mechanism of action and uses.
- c) Explain in brief the life cycle of DNA - virus. Discuss how Indinavir, AZT, Acyclovir and amantidine exhibit their activity.
- d) Explain any one of the following. Mention at least one class of drugs used in its management with mode of action.
  - i) Leprosy
  - ii) Candidiasis

**Q2)** Answer any two of the following. **[10]**

- a) Explain intra and interneuronal signal transmission. Discuss how following classes of drugs affect this process.
  - i) MAO Inhibitors
  - ii) Selective serotonin reuptake inhibitors.
- b) Discuss in brief the role of the following class of compounds in cancer management.
  - i) Alkylating agents
  - ii) Antimitotics
  - iii) Plant products.

**P.T.O.**

- c) Discuss the following in brief. (any two)
  - i) Enzyme inhibitors as drug.
  - ii) Sedatives.
  - iii) Fluoroquinolones.

**Q3) Discuss in brief any three of the following. [6]**

- a) Mechanisms of drug resistance.
- b) Aminoglycosides.
- c) Membrane disruptors as antifungal agents.
- d) Convulsions.

### SECTION - II

**Q4) Explain / Answer any three of the following. [9]**

- a) What are common GIT disorders ? Give the brief overview of the following GIT disorders and at least one drug of choice to treat them.
  - i) Peptic Ulcers
  - ii) Constipation
  - iii) Emesis
- b) Explain the life cycle of plasmodium and explain the role of mefloquin and pyrimethamine as antimalarials and their mechanism of actions.
- c) What is diabetis ? How NIDDM differs from IDDM.Explain how oral hypoglycemic agents control the blood sugar level.
- d) Explain in brief the treatment of following.
  - i) Tuberculosis
  - ii) Pain

**Q5) Answer any two of the following. [10]**

- a) Explain in brief any two of the following cvs disorders. Discuss the pathophysiological changes and at least one drug to treat them.
  - i) Stroke
  - ii) Angina Pectoris
  - iii) Myocardial Infarction.

- b) Explain how following group of compounds help in management of disease (any three)
- i) Cardiac glycosides
  - ii) Selective cox - 2 Inhibitors
  - iii) Ca<sup>2+</sup> blockers
  - iv) Sulphonamides.
- c) How does the endocrine system maintain homeostasis. Explain the role of prolactin, aldosterone, TSH, oxytocin, parathyroid hormone.

**Q6)** Give the mode of action and uses of following drugs. (any three) **[6]**

- a) Rosiglitazone
- b) Pyrazinamide
- c) Cefalexin
- d) Domperidone
- e) Methotrexate
- f) Itraconazole



Total No. of Questions : 6]

SEAT No. :

**P1256**

**[5432]-4003**

[Total No. of Pages : 2

**M.Sc. - II**

**DRUG CHEMISTRY**

**CHD - 463 : Principles and Applications in Drug Design  
(2013 Pattern) (Semester - IV)**

*Time : 3 Hours]*

*[Max. Marks : 50*

*Instructions to the candidates:*

- 1) *All questions are compulsory.*
- 2) *Answers to the two sections should be written in separate answer books.*
- 3) *Figures to the right indicate full marks.*

**SECTION - I**

**Q1) Answer any three of the following :** **[9]**

- a) Explain Hybridoma technology with schematic diagram.
- b) Explain Gene therapy. How it can be used in cancer treatment?
- c) Define the following :
  - i) Affinity.
  - ii) Efficacy.
  - iii) Potency.
- d) What is Antisense technology? How it can be used to treat viral diseases?

**Q2) Answer any three of the following :** **[12]**

- a) Describe signalling mechanism for the tyrosine kinase receptor family.
- b) Discuss the Receptor theories of drug action.
- c) Enlist the various Recombinant DNA produced Medicinal agents or products from the following :
  - i) Enzymes.
  - ii) Vaccines.
- d) Define combinatorial chemistry. Discuss how it will helps to synthesize large number of compounds.

**Q3) Write short notes on any two of the following :** **[4]**

- a) Design of Agonists.
- b) Prodrugs.
- c) Database handling.

***P.T.O.***



## SECTION - II

**Q4)** Answer any three of the following : **[9]**

- a) Discuss in brief of the following :
  - i) Free Wilson method.
  - ii) Topliss scheme.
- b) What is Parallel synthesis? Explain Haughton's tea bag procedure.
- c) Comment on the structure of 7-TM Receptor with schematic diagram.
- d) What is need for prodrug design? Explain with suitable examples.

**Q5)** Answer any three of the following : **[12]**

- a) How are the following are calculated or determined experimentally in QSAR.
  - i)  $E_s$
  - ii) Optimum log P
  - iii)  $\sigma$
  - iv)  $\pi$
- b) Explain mix and split method used in combinatorial chemistry.
- c) Discuss of the following :
  - i) Monte Carlo sampling.
  - ii) Systematic search.
- d) Explain De Novo design method used in designing of molecules when structure is unknown.

**Q6)** Write short notes on any two of the following : **[4]**

- a) Design of Enzyme inhibitors.
- b) Craig plot.
- c) Drug target.



Total No. of Questions : 9]

SEAT No. :

**P1257**

**[5432]-4004**

[Total No. of Pages : 3

**M.Sc. - II**

**DRUG CHEMISTRY**

**CHD - 464 A : Bioinformatics, Cheminformatics and Biostatistics in Drug Discovery and Design**

**CHD - 464 B : Current Trends in Organic Chemistry, Supramolecular, Green Chemistry, Photochemical and Free Radical Reactions**

**CHD - 464 C : Entrepreneurship Development and Project Management (2013 Pattern) (Semester - IV)**

*Time : 3 Hours]*

*[Max. Marks : 50*

*Instructions to the candidates:*

- 1) *Attempt any two of the 464 A, 464 B, 464 C sections only.*
- 2) *Each section is for 25 marks.*
- 3) *All questions are compulsory.*
- 4) *Answer to the two sections to be written in separate answer books.*
- 5) *Figures in right indicate maximum marks.*

**SECTION - I**

(CHD - 464 A : Bioinformatics, Cheminformatics and Biostatistics in Drug Discovery and Design)

**Q1) Answer any three of the following: [12]**

- a) A hen lays eight eggs. Each egg was weighed and recorded as follows. 60, 56, 61, 68, 51, 53, 69, 54.  
Calculate mean, standard deviation.
- b) The percentage of  $\text{CaCO}_3$  content of 10 soil samples is as follows. Find the variance  
10, 8, 7, 8, 4, 9, 12, 13, 16, 14.
- c) The following table shows the amount of diesel required by a vehicle to travel certain distances.

Distance $x$ (km)	90	150	230	310	390
Distance $y$ (Litres)	19.2	33.9	49.0	79.5	89.9

Calculate Karl Pearson correlation coefficient.

*P.T.O.*

- d) Define the following:
- Median
  - Chi-square test
  - Standard deviation
  - Coefficient of variation

**Q2)** Attempt any two of the following: [8]

- Short note on Metabolomics.
- Describe the type of Biological databases.
- “Cheminformatics can be used in drug design”- Justify with example.

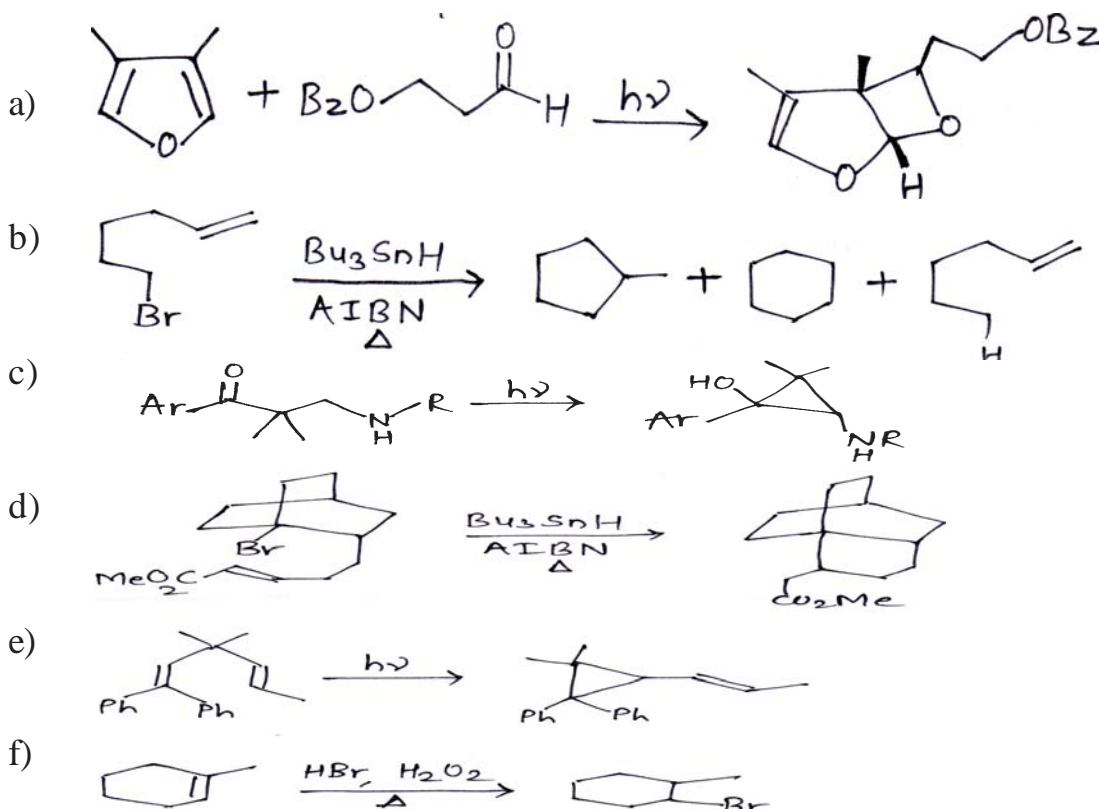
**Q3)** Attempt any two of the following: [5]

- Short note on : Gene prediction programs.
- Give the use and significance of canonical representations in cheminformatics.
- Briefly describe the concept of proteomics.

### SECTION - II

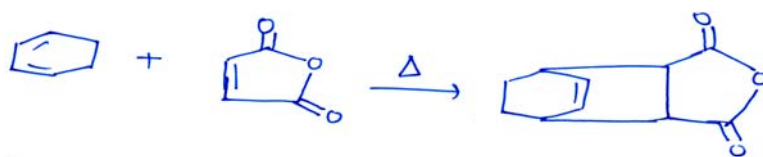
(CHD - 464 B : Current trends in Organic Chemistry : Supra-molecular, Green Chemistry, Photochemistry, and Free Radical Reactions)

**Q4)** Suggest the mechanism and explain the following. (any five) [10]



**Q5)** Solve the following (any four) **[10]**

- a) What are ionic liquids? Give examples of organic reactions involving the use of ionic liquids.
- b) Define atom economy. Calculate atom economy for following reaction.



- c) Give spherical recognition - cryptates of metal ion.
- d) Write the applications of Green Chemistry in organic synthesis.
- e) Make a comment on microwave assisted solvent free reactions with suitable examples.

**Q6)** Answer the following. (any two) **[5]**

- a) What is captodative effect? Explain the factors affecting stability of free radicals.
- b) Photo-Fries rearrangement.
- c) "Cryptands", - make a comment.

### **SECTION - III**

(CHD - 464 C : Entrepreneurship Development and Project Management)

**Q7)** Write short notes on any three of the following: **[6]**

- a) Entrepreneurship Development Process.
- b) Organization and Management.
- c) Corporate Entrepreneurship.
- d) Creativity and Innovation.

**Q8)** Answer any three of the following: **[9]**

- a) 'Entrepreneurship does not emerge spontaneously', Discuss in brief.
- b) Opportunities for small entrepreneurs in India. Explain.
- c) 'Profit is the reward of Entrepreneur'. Comment on the statement.
- d) Differentiate between a manager and Entrepreneur.

**Q9)** Answer any two of the following: **[10]**

- a) Give a brief account of factors affecting entrepreneurial growth.
- b) What are the contents of a business plan? Explain in details.
- c) Explain the problems faced by Women Entrepreneurs.

